

## Synthesis of GLA-60 Positional Isomer as an LPS-Agonist and Its Activity

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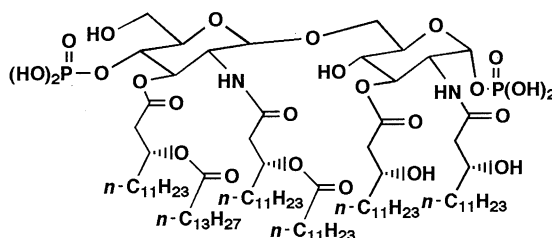
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(Received May 8, 1996)

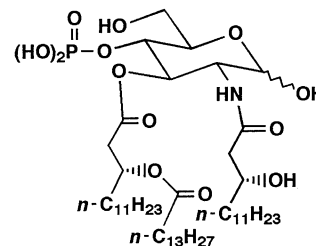
Compound **10** was synthesized from  $\beta$ -D-glucose pentaacetate in a stereocontrolled manner. It unexpectedly showed LPS-agonistic activity much stronger than that of GLA-60 toward human monoblastic U937 cells in the TNF $\alpha$  production.

In spite of the fact that lipopolysaccharide (LPS), an outer surface membrane component in Gram-negative bacteria, causes fever and lethal endotoxic shock in the septicemia of higher animals, it is also known as a highly potent stimulator of the immune system. Most of the biological activities of LPS reside in a relatively small portion of the molecule known as lipid A, a disaccharide unit bearing the constituent lipid moiety. Lipid A, which was first isolated by Westphal and Luderitz<sup>1</sup> and later chemically synthesized by Imoto<sup>2</sup> and Achiwa,<sup>3</sup> exists as a hydrophobic anchor substance holding an essentially linear polysaccharide chain to the cell wall.<sup>4</sup> In a series of structure-activity relationship studies on non-reducing subunit analogues of lipid A, Hasegawa and Kiso<sup>5</sup> have demonstrated that several LPS-agonistic activities are expressed by certain 4-*O*-phosphono-D-glucosamine derivatives pertaining to the structure of GLA-60.<sup>5</sup> Recently it has been shown that lipid A analogues also have LPS-antagonistic activity.<sup>6</sup> Therefore, we have been investigating the compounds related to GLA-60 for therapeutic use. In this paper, we describe the synthesis of GLA-60 positional isomer, *N*-[3-*O*-phosphono-2-*O*-[(*R*)-3-(tetradecanoyloxy)tetradecanoyl]- $\beta$ -D-glucopyranosyl]-3-[(*R*)-hydroxy]tetradecanamide (**10**), which has exhibited LPS-agonistic activity.

The starting  $\beta$ -D-glucose pentaacetate (**1**) was employed for the synthesis of the title compounds. Treatment of **1** in CH<sub>2</sub>Cl<sub>2</sub> with trimethylsilyl azide using tin(IV) chloride as a Lewis acid for 30 min at 25 °C gave an azide **2** (94%) stereoselectively,<sup>7</sup> and the azido group was hydrogenolyzed in THF-H<sub>2</sub>O (2:1) at 25 °C to an amine **3** (44%, after silica gel chromatography) using Pd(OH)<sub>2</sub> on carbon as a catalyst. Silica gel chromatography of the amine **3** may cause the yield low by absorption. Compound **3** was treated with (*R*)-3-(benzyloxy)tetradecanoyl chloride and triethylamine in dichloromethane for 30 min at 25 °C to yield an amide **4** (84%). The coupling constant between C1-H and C2-H of **4** was *J*=9.2 Hz. This reveals that C1-H and C2-H are oriented to  $\alpha$ -axial and  $\beta$ -axial, respectively. Compound **4** was converted to acetamide **5** (72%) by treatment with sodium methoxide in methanol for 30 min at 25 °C, and then with 2,2-dimethoxypropane for 3 h at 25 °C using pyridium *p*-toluenesulphonate (PPTS) as a catalyst in *N,N*-dimethylformamide (DMF) in succession. This 2,3-diol compound **5** was treated with (*R*)-3-(tetradecanoyloxy)tetradecanoic acid,<sup>8</sup> 4-dimethylaminopyridine (DMAP) and dicyclohexylcarbodiimide (DCC) in DMF for 1 h at 25 °C to



Lipid A

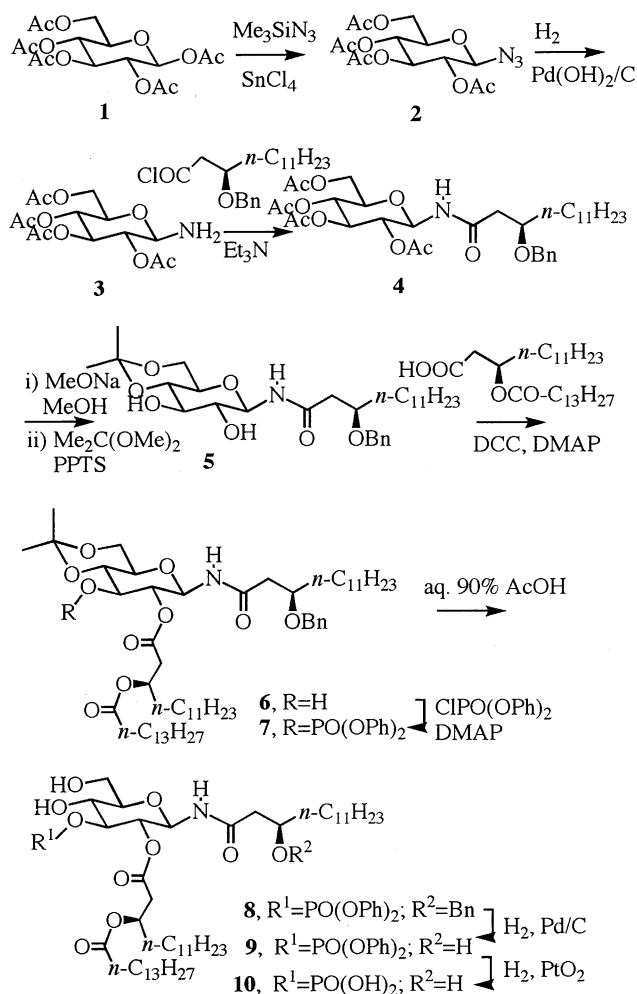


GLA-60

Figure 1. Structures of Lipid A and GLA-60.

give 2-*O*-monoacylated compound **6** (67%) regioselectively. Verification that acylation has occurred at the 2-position of this 2,3-diol compound came from <sup>1</sup>H NMR data of compounds **5** and **6**, showing the C2 proton of compound **6** had shifted to lower magnetic fields than that of compound **5** (from  $\delta$  3.20 to 4.70). The remaining 3-hydroxy group of **6** was phosphorylated with diphenyl chlorophosphate and DMAP in CH<sub>2</sub>Cl<sub>2</sub> for 30 min at 25 °C to give **7** (94%). The acetonide group of compound **7** was cleaved by treatment with aqueous 90% acetic acid for 5 h at 60 °C to give 4,6-diol **8** (76%). Compound **8** in turn was debenzylated using 10% palladium on carbon in THF for 3 h at 25 °C to give **9** (77%). Finally, hydrogenolysis of phosphate ester **9** in THF for 3-16 h at 25 °C using platinum(IV) oxide as a catalyst yielded the phosphono compound **10** (99%).<sup>9</sup> Thus compound **10** was synthesized in a stereocontrolled manner.

Compound **10** unexpectedly showed LPS-agonistic activity much stronger than that of GLA-60 toward human monoblastic U937 cells. The TNF $\alpha$  production (% of control; 10 ng/ml of LPS=100) of lipid A, GLA-60, and compound **10** in the concentrations of 0.01  $\mu$ M, 0.1  $\mu$ M, 1  $\mu$ M, and 10  $\mu$ M were as follows. Lipid A was 21.0, 136.0, 385.0, and 651.0; GLA-60



Scheme 1.

was 12.0, 12.0, 7.0, and 13.0; and compound **10** was 14.0, 20.0, 57.0, and 611.0, respectively. Comparing the structure of **10** with GLA-60 or lipid A, the positions of the substituents were each shifted one position. Moreover, the substitutions at the C1 (anomeric amido), C2 (ester), and C3 (phosphono) positions of compound **10** are aligned with C1- $\beta$ , C2- $\alpha$ , and C3- $\beta$ , respectively, and those of GLA-60 and lipid A are aligned with C2- $\alpha$  (amido), C3- $\beta$  (ester), and C4- $\alpha$  (phosphono), respectively. Nevertheless, U937 cell recognized **10** as an LPS-agonist just like GLA-60 and lipid A. We cannot provide a reasonable explanation for this phenomenon. This finding may offer a

new concept for less toxic anticancer drugs relating to GLA-60 and lipid A.

## References and Notes

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- a) Commercially available (*R*)-3-hydroxytetradecanoic acid was converted to (*R*)-3-tetradecanoyloxytetradecanoic acid in three steps as follows: (i) esterification of the acid with diphenyl diazomethane, (ii) acylation of the 3-hydroxy group with tetradecanoyl chloride and triethyl amine, and (iii) deprotection of the diphenylmethyl ester with trifluoroacetic acid or by hydrogenolysis. b) cf. An alternative route. P. K. Jadhav, *Tetrahedron Lett.*, **30**, 4763 (1989).
- <sup>1</sup>H NMR (pyridine-d<sub>5</sub>) of **10**:  $\delta$  0.89 (9H, t,  $J=6.6$  Hz), 1.05-1.60 (56H, m), 1.60-1.82 (6H, m), 2.51 (2H, t,  $J=7.3$  Hz), 2.77 (1H, dd,  $J=4.4, 14.7$  Hz), 2.84 (1H, dd,  $J=6.6, 14.7$  Hz), 2.95 (1H, dd,  $J=7.3, 16.1$  Hz), 3.05 (1H, dd,  $J=7.3, 16.1$  Hz), 3.93-4.00 (1H, m, C<sub>5</sub>-H), 4.20-4.52 [4H, m, C<sub>4</sub>-H, C<sub>6</sub>-H<sub>2</sub>, CH-(OH)-C<sub>11</sub>H<sub>23</sub>], 5.12 (1H, q,  $J_{2,3}=J_{3,4}=J_{\text{HCOP}}=8.8$  Hz, C<sub>3</sub>-H), 5.57-5.68 [2H, m, C<sub>2</sub>-H, CH-(OCOC<sub>13</sub>H<sub>27</sub>)-C<sub>11</sub>H<sub>23</sub>], 6.10 (1H, t,  $J=8.8-9.4$  Hz, changed to a doublet,  $J=9.4$  Hz, on addition of D<sub>2</sub>O, C<sub>1</sub>-H), 6.21 (5H, broad, OH x 5), 9.81 (1H, d,  $J=8.8$  Hz, C<sub>1</sub>-NH).